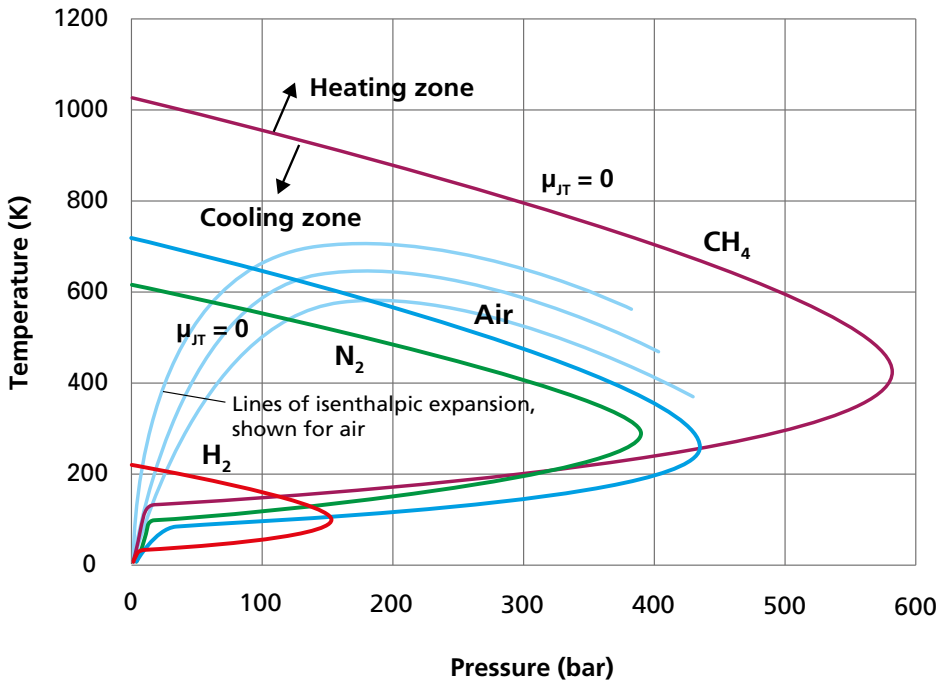
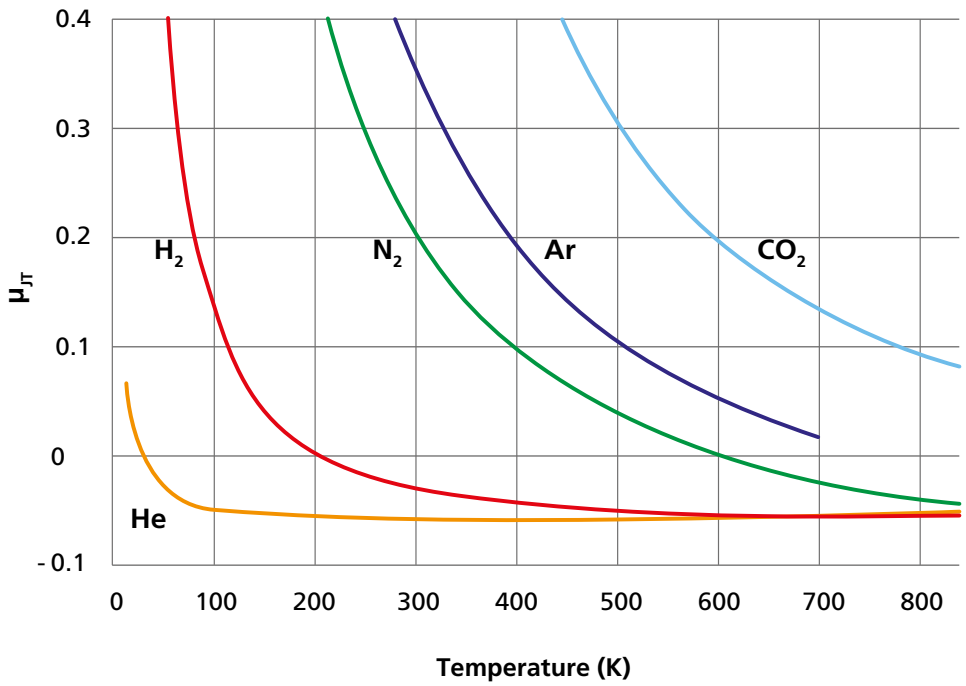
Joule-Thomson Effect for Heating and Cooling of Gases as They Expand

Joule-Thomson Inversion Curve



μ_{JT} – Joule-Thomson coefficient (K/bar) and inversion temperatures at atmospheric pressure



H_2 Hydrogen
N_2 Nitrogen
CH_4 Methane
He Helium
Ar Argon
CO_2 Carbon dioxide

$\mu_{JT} > 0$ means the gas will cool down during expansion
$\mu_{JT} < 0$ means the gas will warm up during expansion

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