

# Driving Efficiency: Energy Optimisation in CCU Processes

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sbh4 GmbH, Stephen B. Harrison  
12<sup>th</sup> European Carbon Dioxide Utilisation Summit  
Antwerp, 25<sup>th</sup> September 2025  
10:35 to 11:05

# Agenda

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## Driving Efficiency: Energy Optimisation in CCU Processes

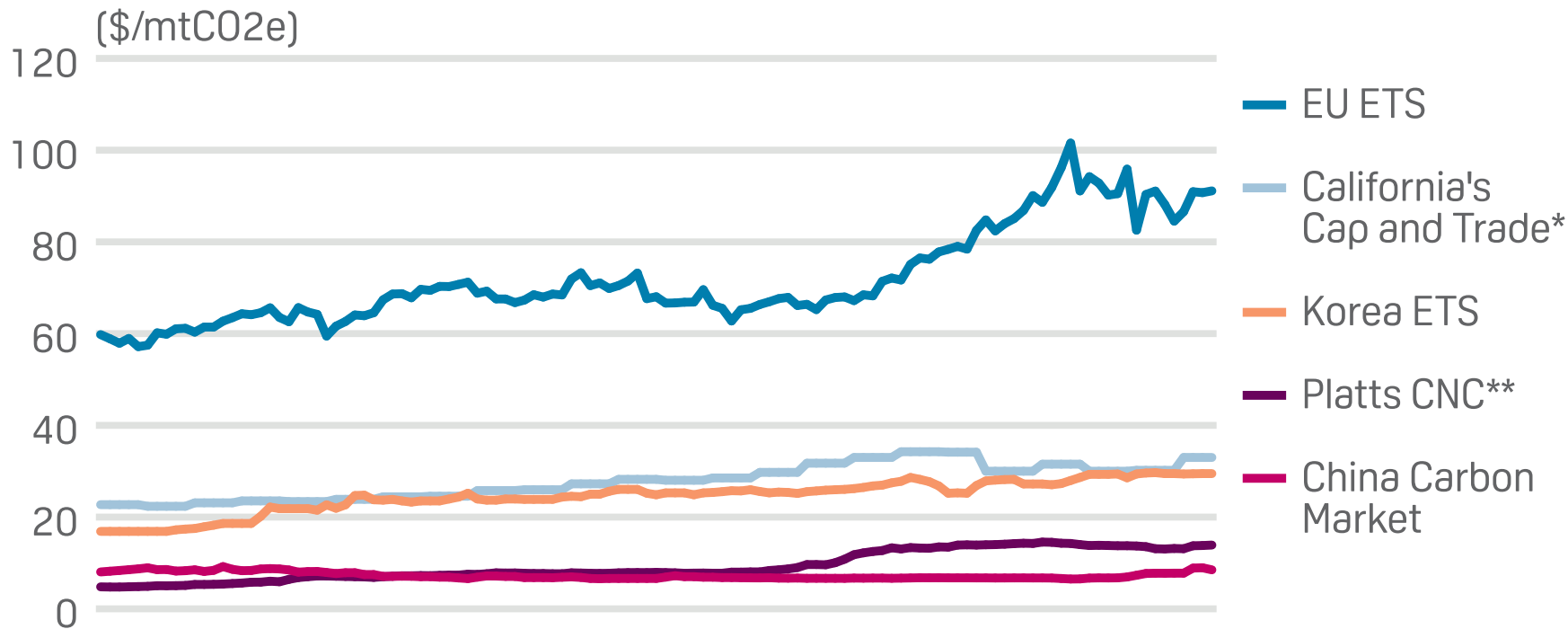
1. Key social, economic, and policy factors that will shape the future landscape of the industry
2. Balancing operational efficiency with emissions reduction to achieve net-positive environmental impacts
3. Emerging innovations that could revolutionise CCU capabilities and applications

# 1. Key social, economic, and policy factors that will shape the future landscape of the industry

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Policy and CO2 emissions costs drive the business case for CO2 capture for CO2 utilisation

CO2 emissions costs, or opportunity costs around the world are progressively rising and geographic coverage is increasing.



# Two views of CO2 capture at Holcim, Brevic. Social acceptance was and remains key.



# 1.1 Processes that require no (additional) energy for CO<sub>2</sub> capture

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**Lowest incremental capex and opex, brownfield sites, minimal disruption.**

Best economics.

Best social acceptance?

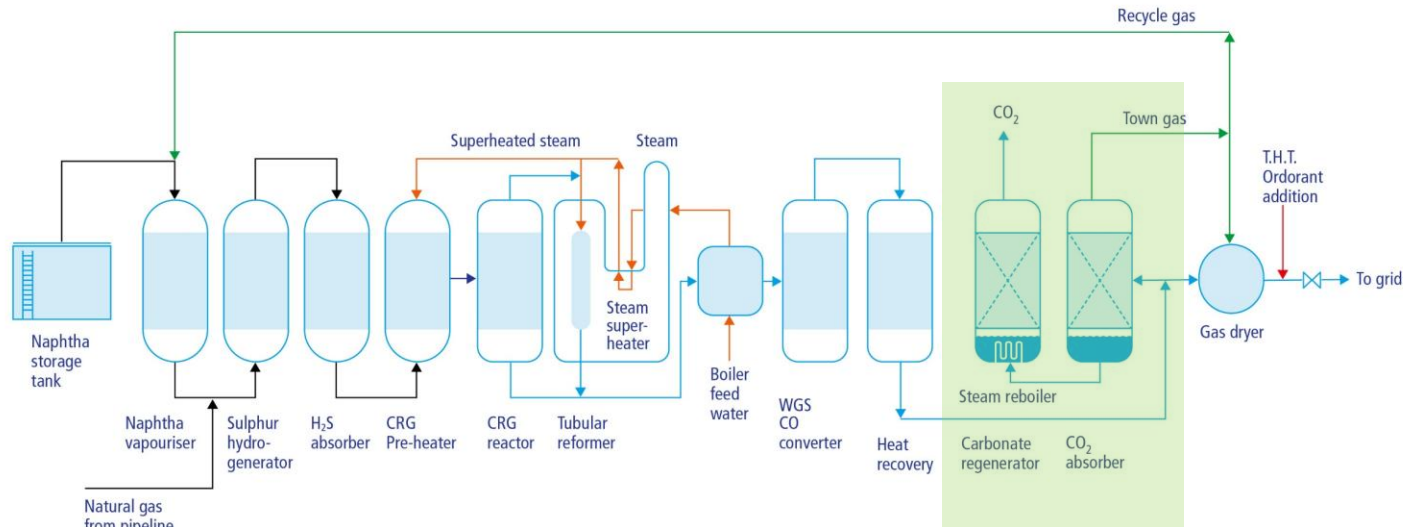
Best policy support?



## Syngas, Towngas – 3,500km of pipelines in Hong Kong for 1.9 million domestic, commercial and industrial customers



# Hong Kong Town Gas – Tai Po Catalytic Rich Gas naphtha / methane reformer and CO<sub>2</sub> capture process



## Town gas composition

Carbon Dioxide	16.3% – 19.9%
Carbon Monoxide	1.0% – 3.1%
Methane	28.2% – 30.7%
Hydrogen	46.3% – 51.8%
Nitrogen and Oxygen	0% – 3.3%

## Town gas energy value

Calorific Value	17.27 MJ/m <sup>3</sup>
Specific Gravity	0.52
Wobbe Index	24

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Hong Kong Town gas consulting

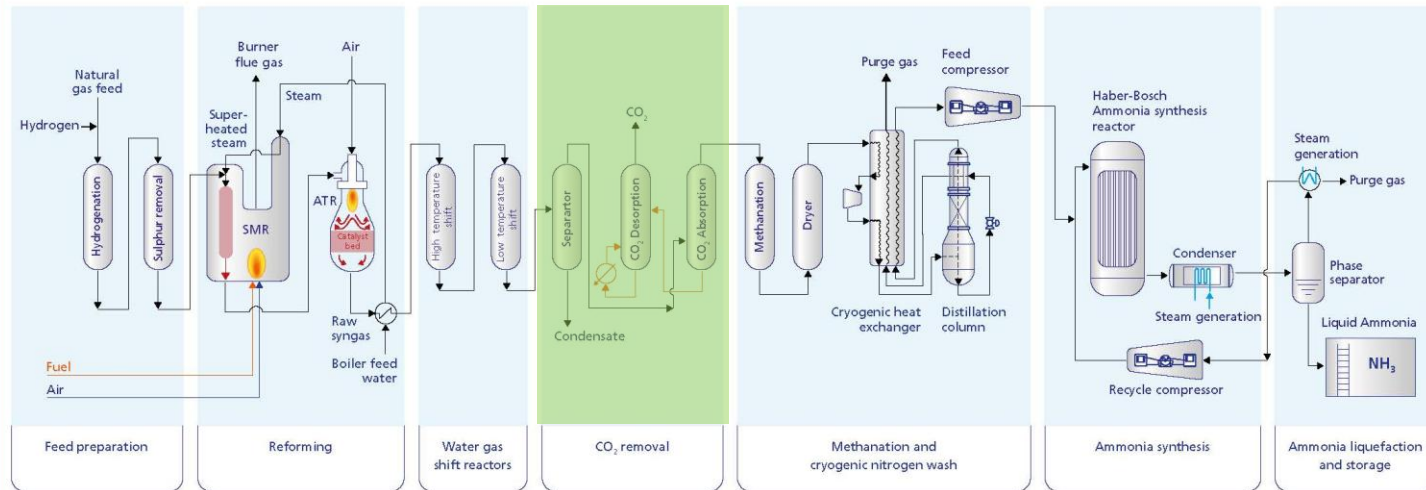
- Cooking and heating fuel
- More than 1 million domestic and commercial customers
- Naphtha reforming on SMR
- CO<sub>2</sub> captured to increase the calorific value of the town gas
- High purity (>95%) CO<sub>2</sub> vented to atmosphere



# CO2 capture and removal (with vent to atmosphere) is integrated into ammonia production.



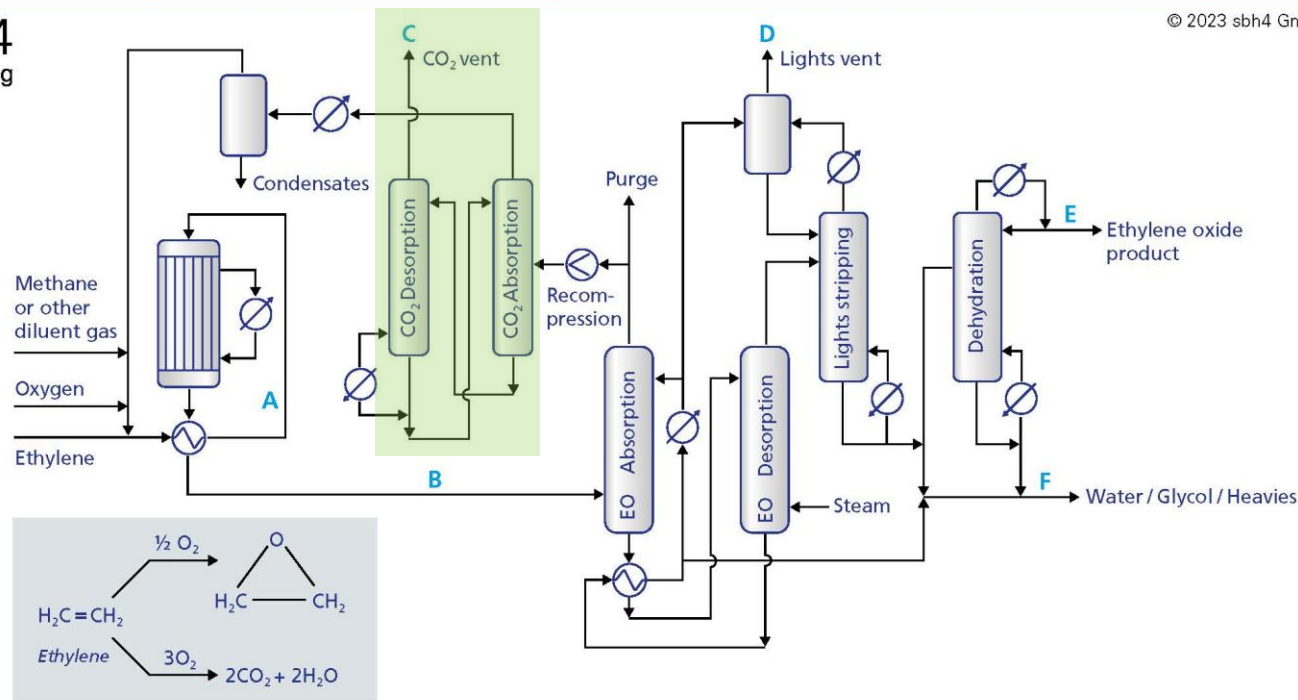
# Air-fed Ammonia Production Process



## Grey ammonia

- The reforming stages for grey ammonia can either be air- or oxygen-fed
- Use of an SMR is with the subsequent use of an ATR is common
- If the ATR is oxygen-fed, pure nitrogen must be added in the ammonia synthesis loop
- Pure oxygen is introduced to the ATR to make hydrogen
- CO<sub>2</sub> emissions from reforming are removed to protect the ammonia synthesis catalyst
- Most of the CO<sub>2</sub> is vented to atmosphere
- Some captured CO<sub>2</sub> is used to make urea or used for food and beverage applications
- Alternatively, CO<sub>2</sub> can be sequestered

# Oxygen-fed Ethylene Oxide Production with Integrated CO<sub>2</sub> Capture



## Ammonia is not unique...

- Ethylene oxide production also captures CO<sub>2</sub> from within the process
- Some EO plants have been accessed as a low-cost source of CO<sub>2</sub> for commercial applications
- As with ammonia, the CO<sub>2</sub> capture costs are taken by the EO product.
- The incremental capex and opex to transition from grey EO to reduced CO<sub>2</sub> EO is purification (drying in this case) and liquefaction or compression and sequestration.

	A: Reactor feed	B: Reactor outlet	C: CO <sub>2</sub> vent	D: Lights vent	E: EO Product	F: Water
Ethylene	34.6 %	25.4 %	0.4 %	64.6 %		
Ethylene oxide (EO)		2.1 %			99.7 %	5.8 %
Oxygen	23.4 %	3.9 %		0.8 %		
Methane	34.6 %	57.2 %	0.4 %	11.0 %		
Water	4.2 %	3.7 %	1.7 %		0.2 %	94.2 %
Carbon Dioxide (CO <sub>2</sub> )	3.2 %	7.7 %	97.5 %	23.6 %	0.1 %	

## 2. Balancing operational efficiency with emissions reduction to achieve net-positive environmental impacts.

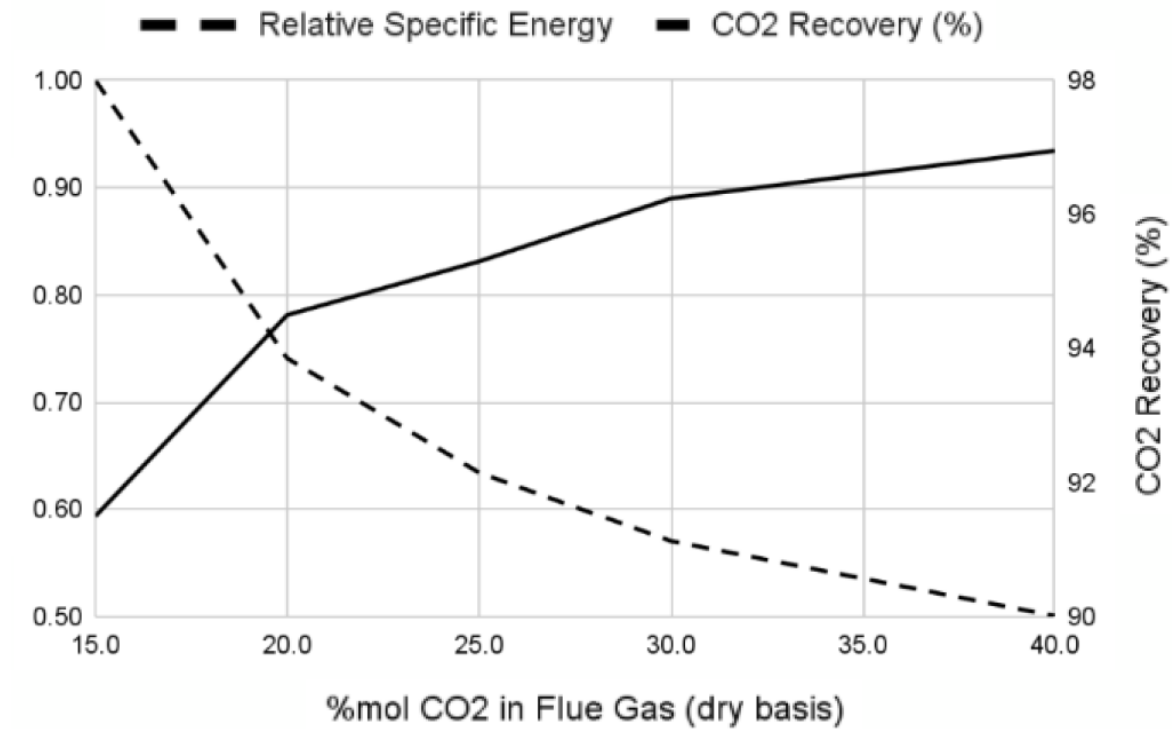
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You can't fight science, but you can choose the right science to work for you.

Technology selection must fit the flue gas input and required CO2 output.

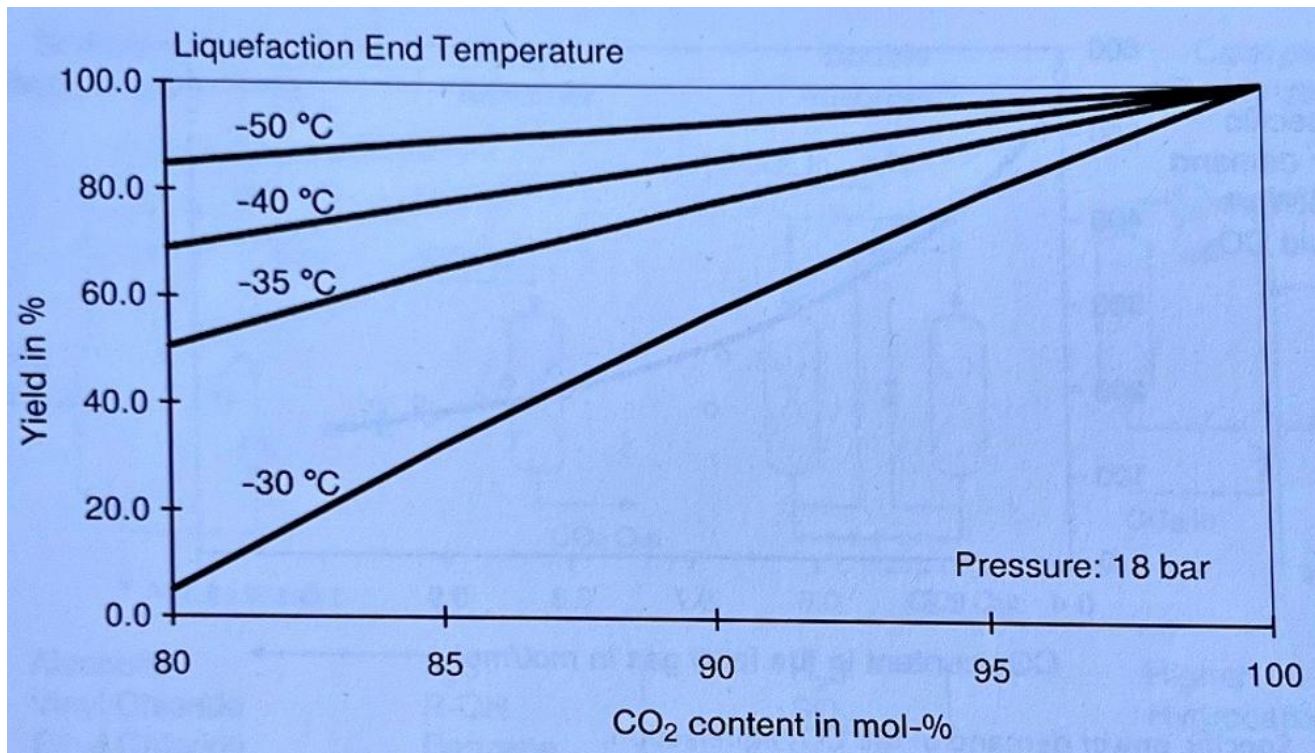
The energy requirements are driven by fundamental thermodynamics and the relevant heat and mass balances for the chosen technology.

For each technology, the specific energy requirement falls as the flue gas CO<sub>2</sub> concentration increases. A high CO<sub>2</sub> capture rate also increases the specific energy requirement.



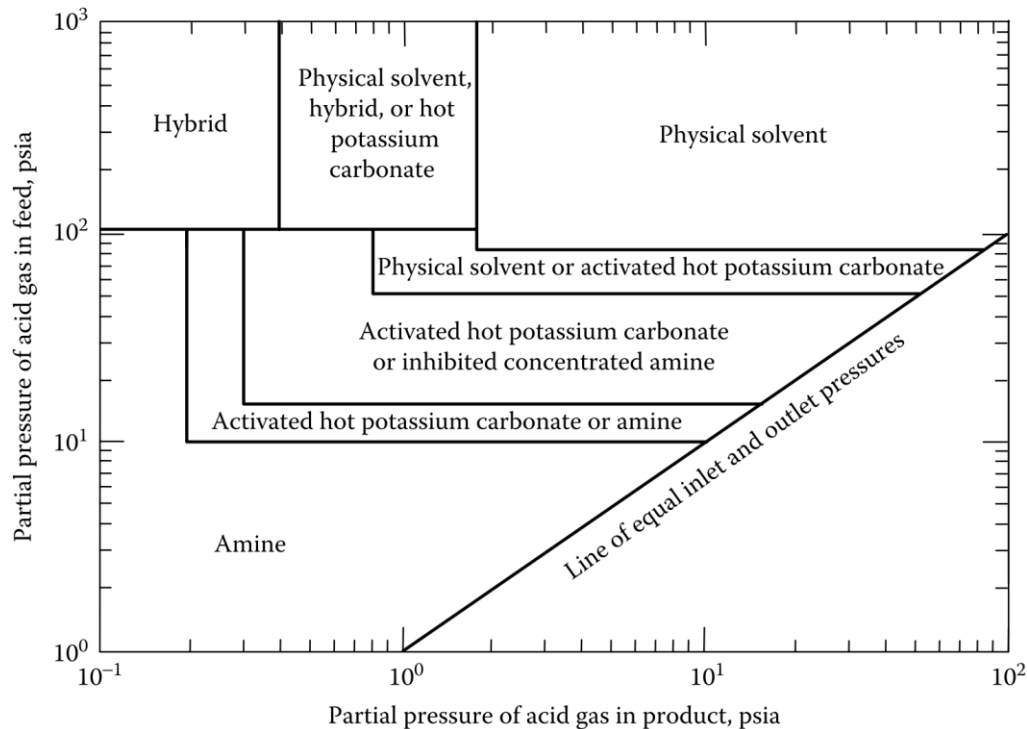


High purity CO<sub>2</sub> feed to the liquefier results in less CO<sub>2</sub> losses from the liquefier inert gas (N<sub>2</sub>/CH<sub>4</sub>/O<sub>2</sub>/CO) purge. Also (not shown on this diagram) dry CO<sub>2</sub> to the liquefier results in less CO<sub>2</sub> losses on the liquefier dryer unit. The holistic system CO<sub>2</sub> losses, or CO<sub>2</sub> capture rate must be analysed.





The CO<sub>2</sub> capture system inlet and outlet partial pressures of CO<sub>2</sub> are important factors for technology selection and within each technology, heavily influence the process efficiency.



Chemical solvents (amine and HPC) react with the CO<sub>2</sub> (chemisorption) and require high regeneration energies.

Physical solvents use physical absorption (physisorption) based on Henry's law. They require high partial pressure CO<sub>2</sub> but benefit from reduced regeneration energy.

- Selexol (Honeywell UOP, Glycol / PEG)
- Rectisol (Linde / AL, methanol)
- Purisol (AL (Lurgi), NMP)
- Sulfinol (Shell, Sulfolane and amines)

This is analogous to solid adsorption systems where MOFs rely on physisorption, and solid amines rely on chemisorption.

### 3. Emerging innovations that could revolutionise CCU capabilities and applications

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To revolutionise CCU, we must cut the cost of CO<sub>2</sub> capture.

Leveraging the right science for each application.

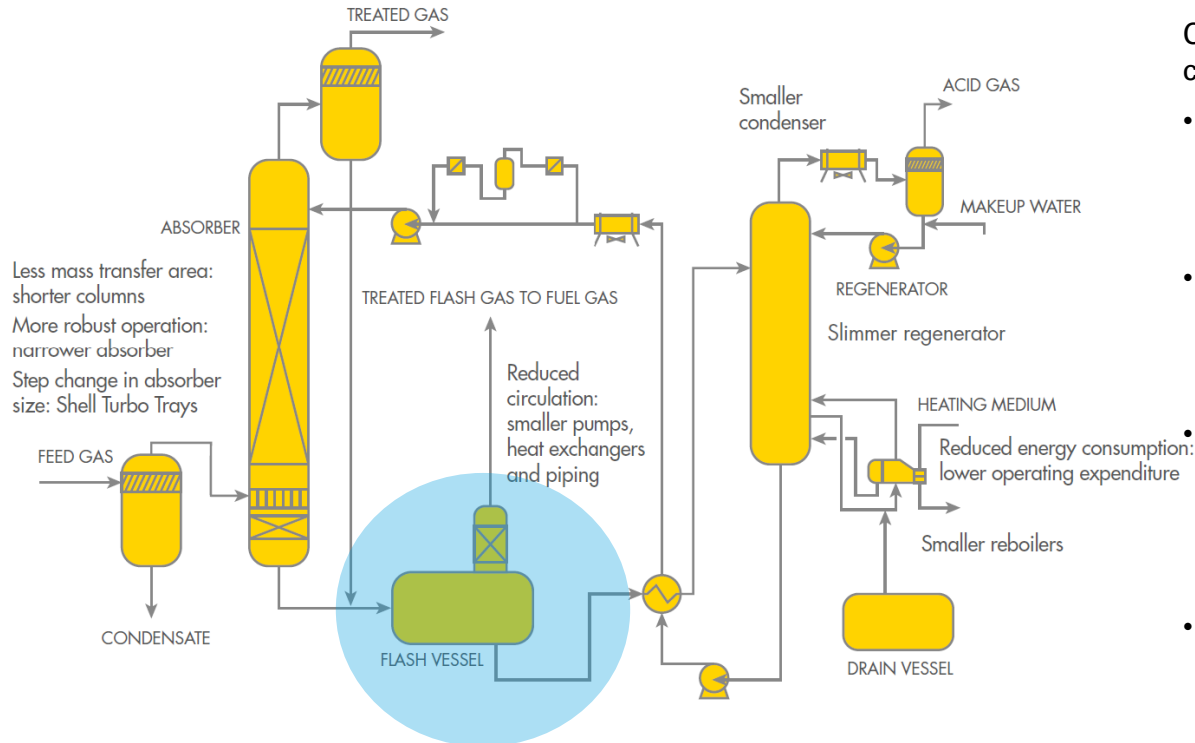
A review of established, emerging and disruptive technologies.

## 3.1 Chemisorption or physisorption in solvent-based CO<sub>2</sub> capture systems

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Physisorption works well with high partial pressure of CO<sub>2</sub> and has a lower regeneration energy since CO<sub>2</sub> is loosely bound to the adsorbent

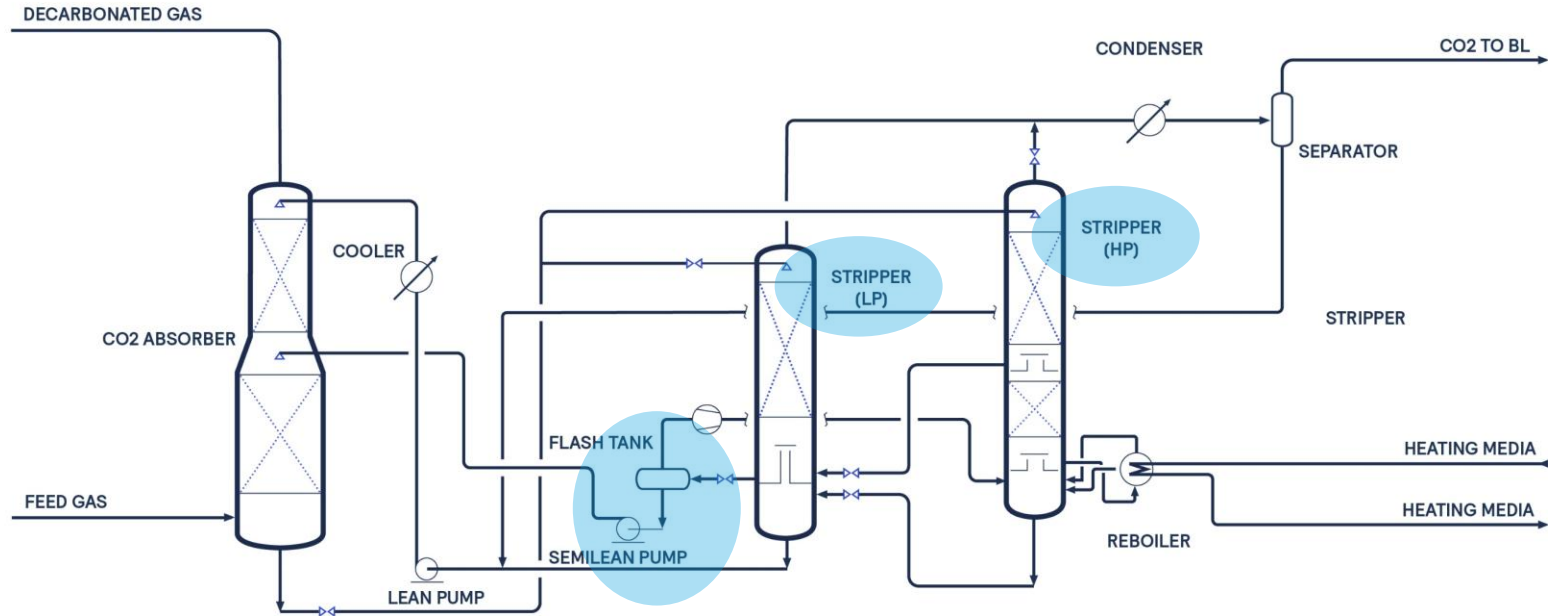
Shell ADIP ULTRA, as implemented at Shell Quest for pre-combustion CCS from SMRs. Amine solvent with a flash between the absorption and stripping columns utilises SMR process gas pressure to reduce regeneration energy.



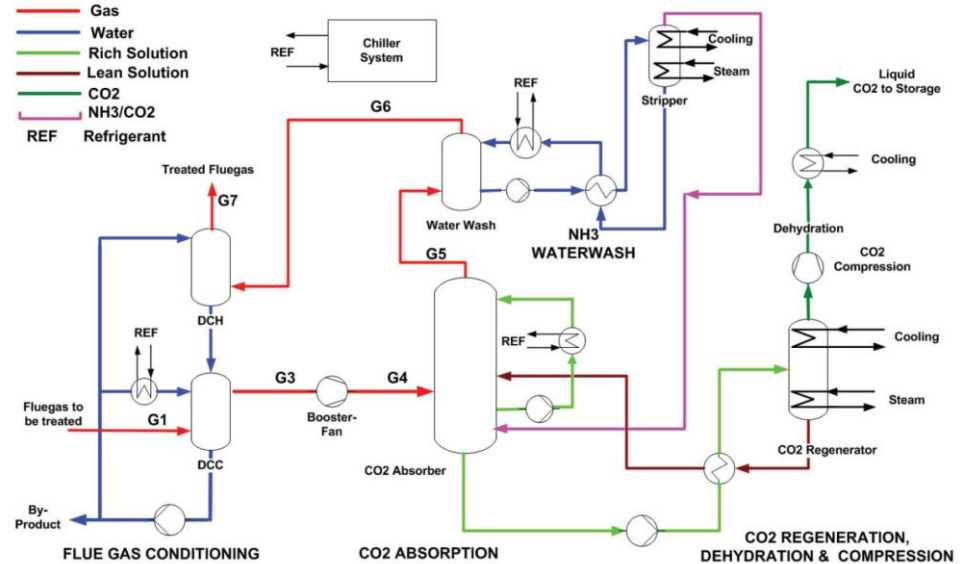
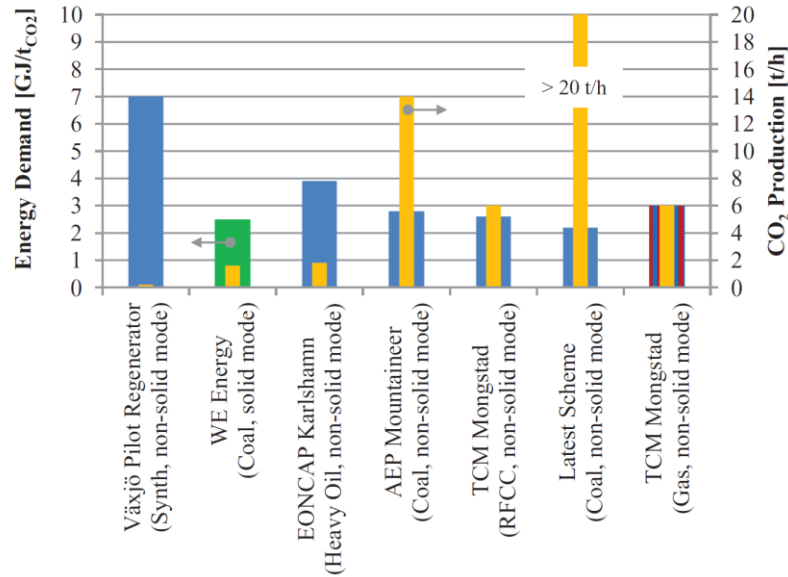
CO<sub>2</sub> capture at pressure has multiple capex and opex benefits

- The CO<sub>2</sub> flash in the high pressure CO<sub>2</sub> capture process releases CO<sub>2</sub> and reduces the heating energy requirement by about 30%.
- High pressure operation means that the solvent regeneration temperature is lower, meaning that lower grade steam can be used.
- The high pressure system is smaller and despite being a pressurised system, capex is reduced by about 30% compared to the very large low-pressure system.
- Footprint is significantly reduced.

Giammarco Technologies NovaFlash®: high-efficiency Hot Potassium Carbonate CO<sub>2</sub> capture for high-pressure streams. A flash between the high-pressure and low-pressure stripper columns, and an additional flash tank reduce the energy input requirement.

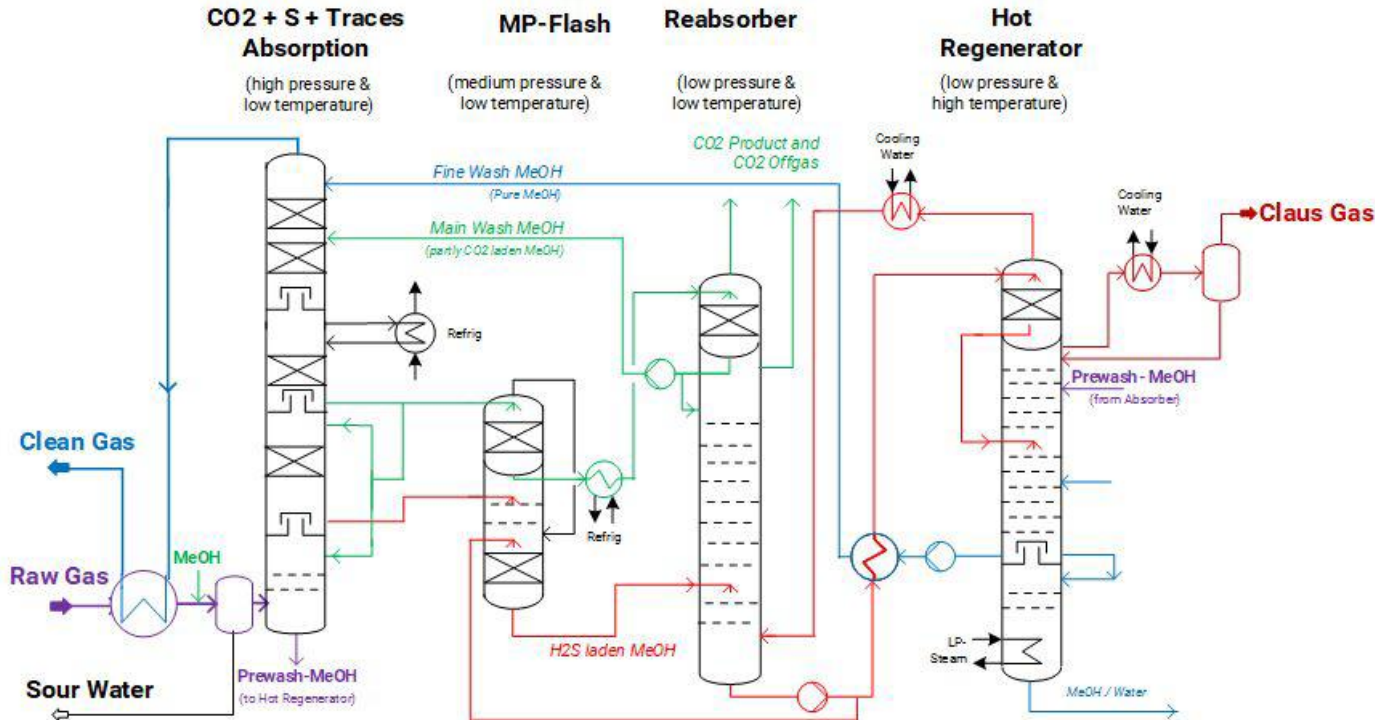


Baker Hughes chilled ammonia process (CAP). Previously developed by GE / Alstom. CAP has been demonstrated at various scales. It can either precipitate ammonium bicarbonate or operate fully in the liquid phase. Chilling reduces ammonia carry over with the flue gas vent. Energy demand is in the range 2 to 3 GJ/tonne CO<sub>2</sub>, like amines.





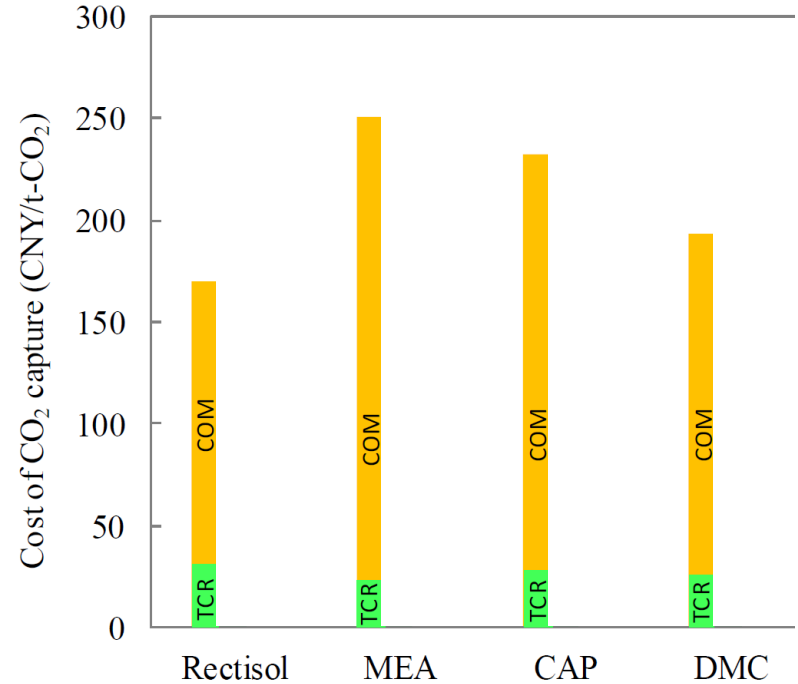
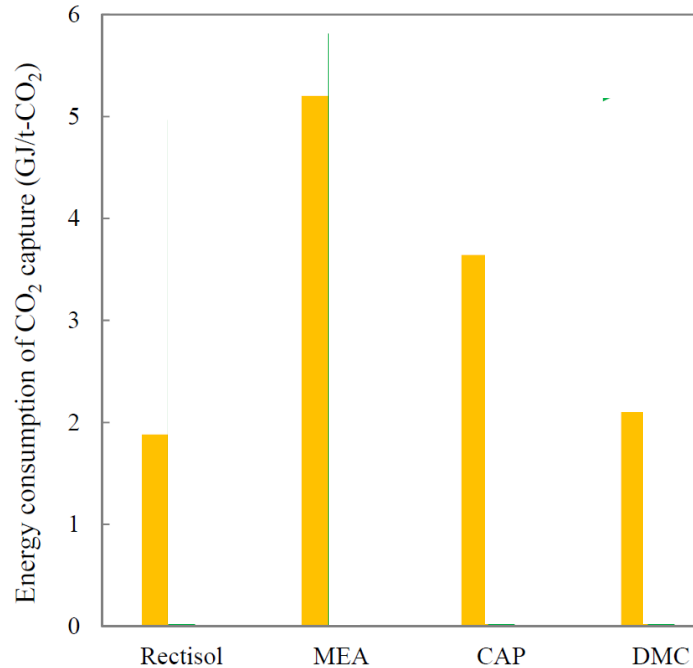
Air Liquide Rectisol™ recovers sulphur chemicals, ammonia, humidity and CO<sub>2</sub> from the syngas in different stages. Heavy metals can also be removed.



Linde (50 references globally) and Air Liquide / Lurgi (110 references globally) offer Rectisol™ using a chilled methanol solvent. The main energy input is compression energy for the methanol chiller circuit.



At large scale, Rectisol™ can offer a lower total cost of ownership (capex, opex and maintenance) than MEA (Amine), CAP (Chilled Ammonia Process) and DMC (Di-methyl carbonate).



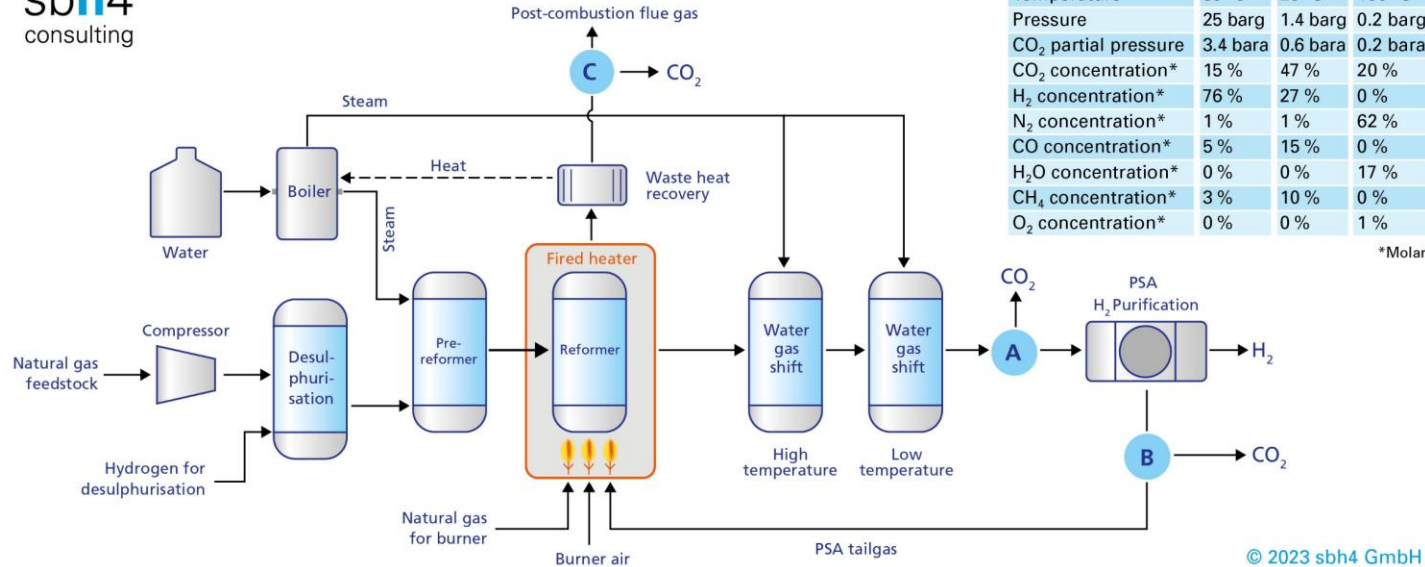
TCR: Total Capital Requirement. COM: Cost of Operation and Maintenance.

## 3.2 Physisorption in solid adsorbent CO<sub>2</sub> capture systems

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Physisorption works well with high partial pressure of CO<sub>2</sub> and has a lower regeneration energy since CO<sub>2</sub> is loosely bound to the adsorbent

# Potential Locations for CO<sub>2</sub> Capture from Steam Methane Reforming



## Retrofit existing SMRs

- Circa 2,000 SMRs operate worldwide from 10 to 500 TPD H<sub>2</sub>
- Existing SMRs can be decarbonised with CO<sub>2</sub> capture retrofits.
- Location of CO<sub>2</sub> capture influences unit cost and efficiency of CO<sub>2</sub> capture
  - Stream pressure changes
  - Stream CO<sub>2</sub> composition changes
  - Maximum potential CO<sub>2</sub> capture rate changes

	Location A	Location B	Location C
Process stage	Pre-PSA	Post-PSA	Post-combustion
Advantages	High pressure, high CO <sub>2</sub> concentration, highest CO <sub>2</sub> partial pressure, lowest unit cost of CO <sub>2</sub> capture from Amine Solvent or VSA processes	Low flowrate (H <sub>2</sub> removed), highest CO <sub>2</sub> concentration	More than 90 % capture rate possible (captures process CO <sub>2</sub> and burner CO <sub>2</sub> emissions), low pressure location can be suitable for emerging CO <sub>2</sub> capture technologies such as TSA and mineralisation
Disadvantages	Max 70 % CO <sub>2</sub> capture rate possible (burner CO <sub>2</sub> emissions not captured), high flowrate (H <sub>2</sub> included)	Max 70 % CO <sub>2</sub> capture rate possible (burner CO <sub>2</sub> emissions not captured), low pressure	Low pressure, lowest CO <sub>2</sub> concentration, high flowrate due to combustion air, highest unit cost of CO <sub>2</sub> capture from Amine Solvent or VSA processes

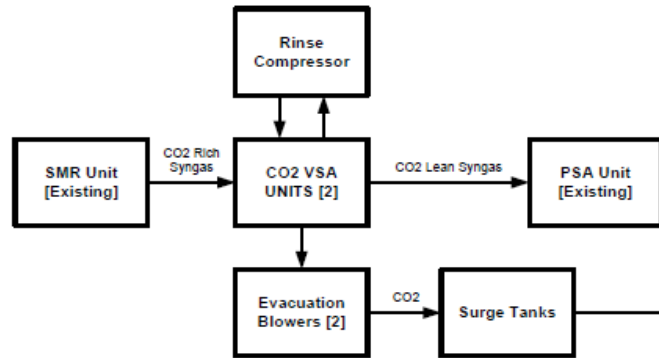


# VSA (Vacuum swing adsorption) with molecular sieve CO<sub>2</sub> adsorbents - proven for carbon capture at two Air Products SMRs feeding the Valero refinery in Port Arthur, USA.



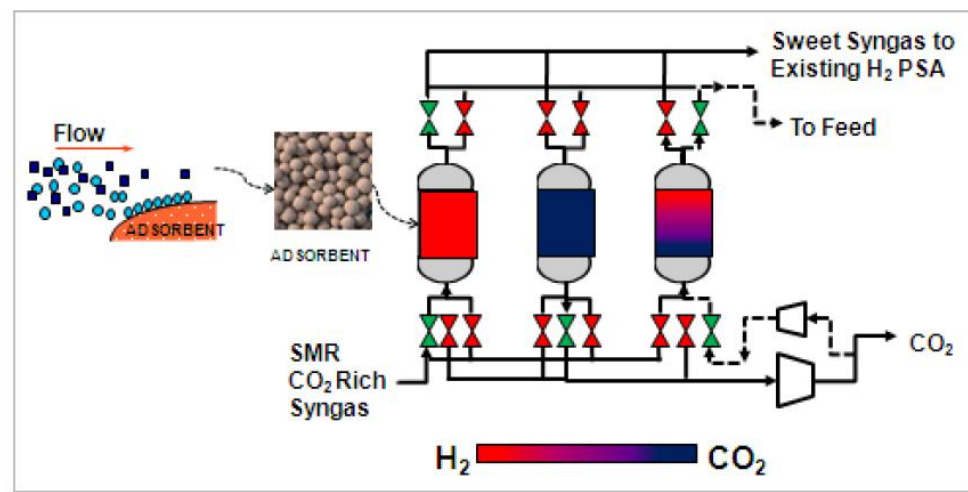
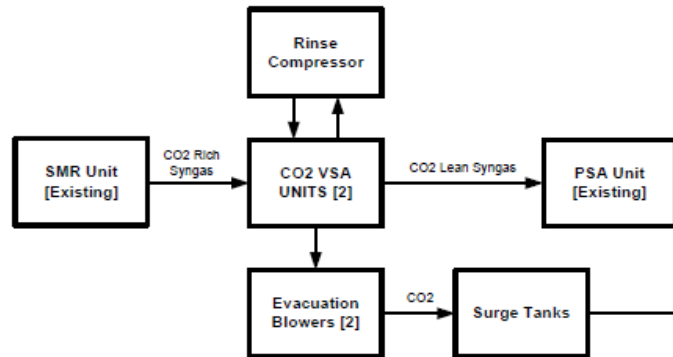


# Air Products VSA and CO2 compression process layout



PORT ARTHUR #1 SITE

PORT ARTHUR #2 SITE



Demonstration of Carbon Capture and Sequestration of Steam Methane Reforming Process Gas Used for Large-Scale Hydrogen Production

## FINAL REPORT

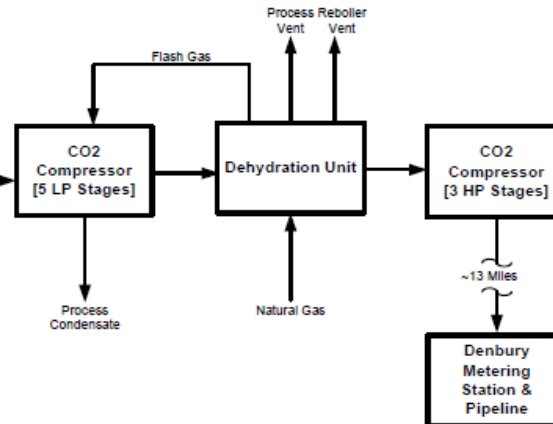
Reporting Period:  
September 2009 – September 2017

Principal Author(s):  
Amy Busse  
Gloria Power  
Joel MacMurray

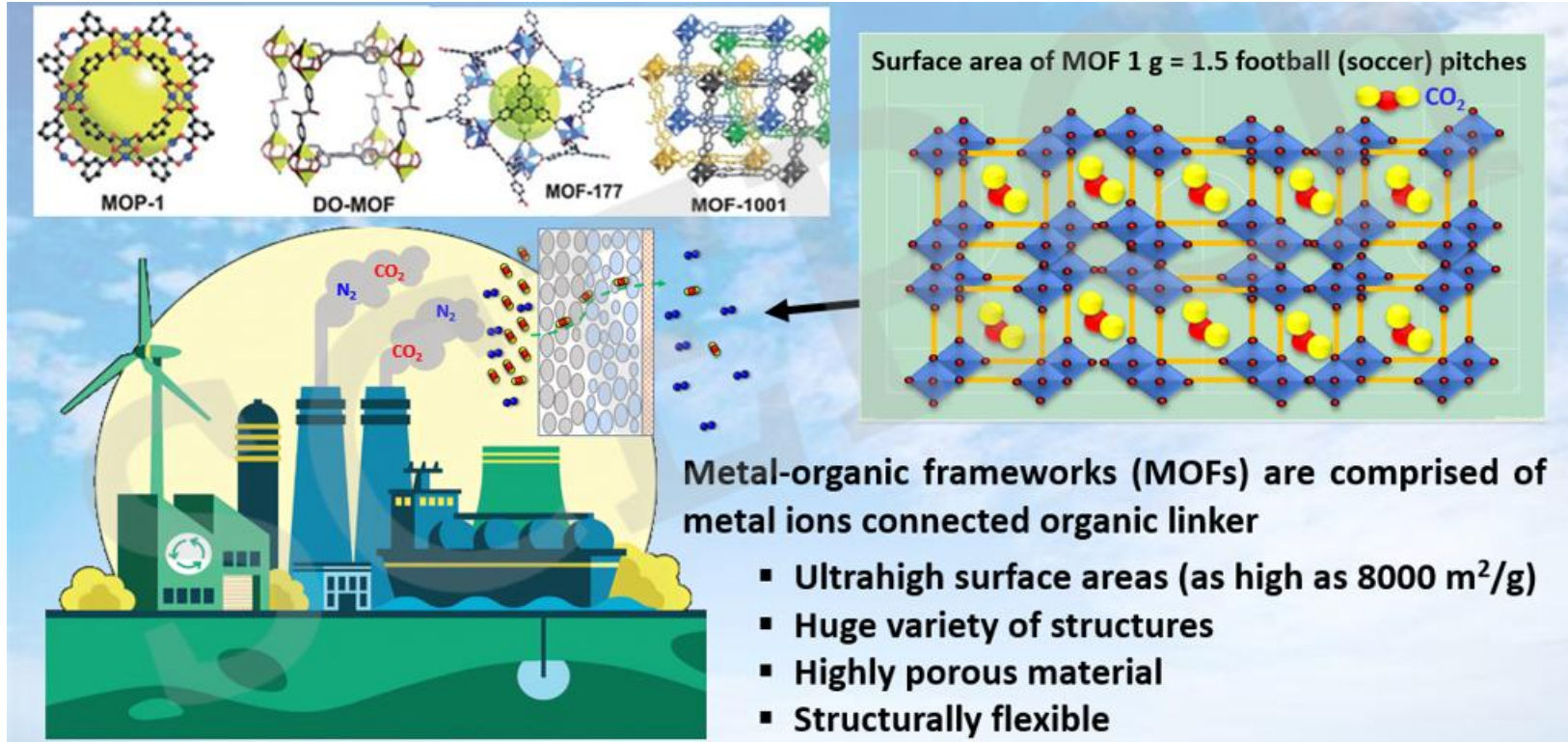
Date Report Issued:  
March 2018

DE-FE0002381

Air Products and Chemicals, Inc.  
7201 Hamilton Blvd.  
Allentown, PA 18195



MOFs can operate in a similar process to VSA with molecular sieves. They have the advantage of using physisorption of CO<sub>2</sub> which reduces the regeneration energy requirement.



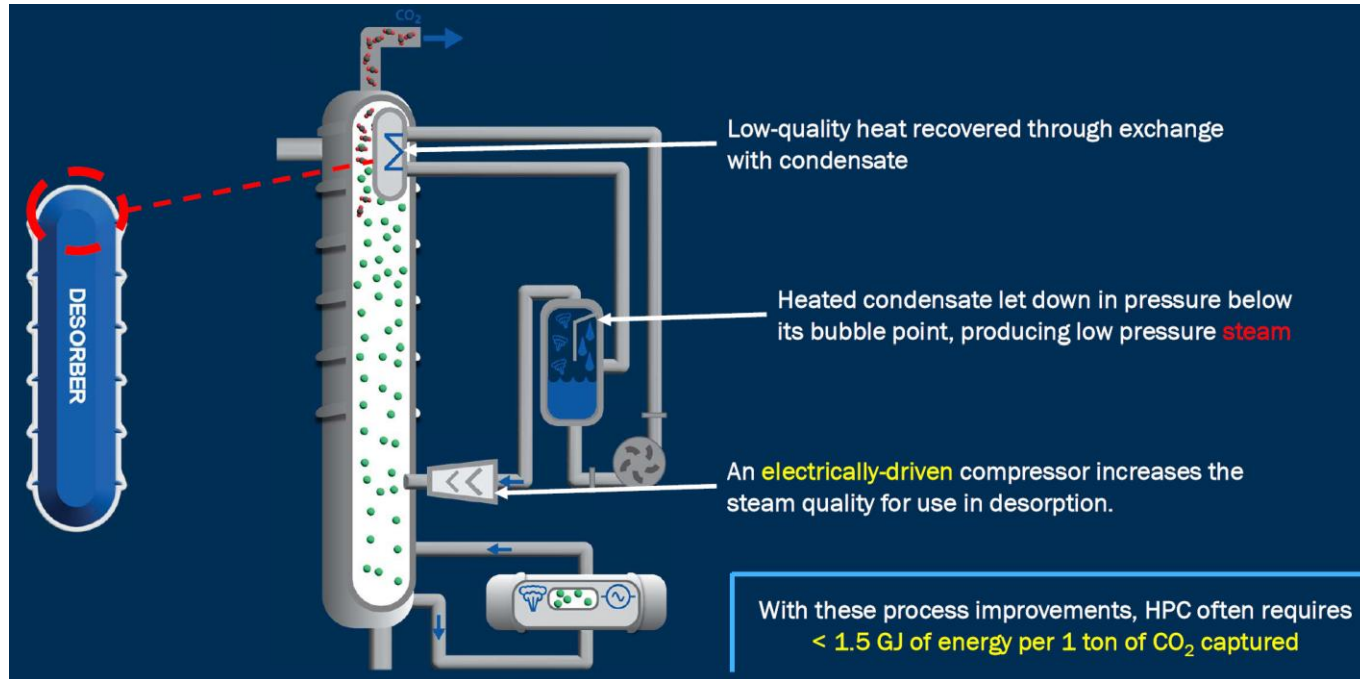
### 3.3 Solvent regeneration energy from steam (dewpoint swing), heat (temperature swing) or electrical power (pressure swing)?

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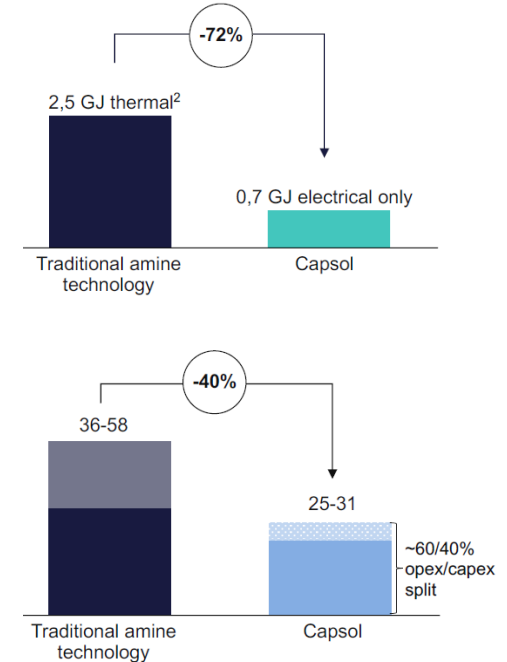
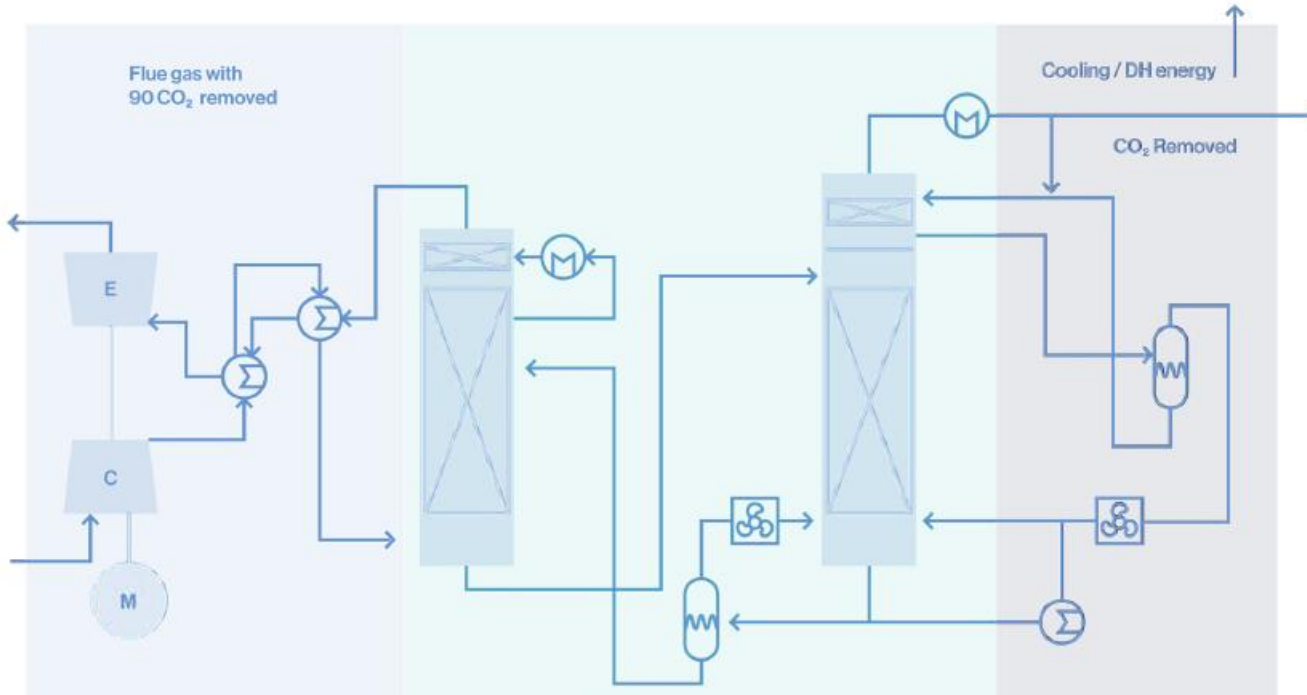
Which energy source is most freely available at the site?

Which can be more easily decarbonised in the future?

Partial electrification to use MVR (mechanical vapour recompression: a type of heat pump). The energy input to the HPC process can be reduced to less than 1.5 GJ energy per 1,000 tonnes of CO<sub>2</sub> captured.



Full electrification: CapsolEoP™ claims 0.7 to 1.5 GJ Electrical power required per 1,000 tonnes of CO<sub>2</sub> captured. The compressor / expander (componder) is key to operating the high pressure HPC process with a flash and recovering pressure energy from the flue gas.



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# Introduction to Stephen B. Harrison and sbh4 consulting

**Stephen B. Harrison** is the founder and managing director at sbh4 GmbH in Germany. His work focuses on decarbonisation and GHG emissions reduction. E-fuels, hydrogen, ammonia and CCTUS are fundamental pillars of his consulting practice.

Stephen has extensive M&A and investment due diligence advisory experience in the energy and clean-tech sectors. Private Equity firms, investment fund managers and green-tech start-ups are regular clients. He also supports operating companies in their mission to decarbonise their scope 1, 2 and 3 GHG emissions.

In 2023, Stephen evaluated seven CCTUS, hydrogen and e-fuels submissions to the European Commission's Third Innovation Fund. The fund allocated €2 billion to large-scale decarbonisation projects in Europe. In 2024 he supported the European Commission with venture capital investment due diligence and assessed eight Horizon grant applications. Again in 2025, Stephen is assessing seven Innovation Fund applications related to e- and bio-methanol production.

Stephen has served as the international expert and team leader for three ADB projects related to CCTUS and renewable hydrogen deployment in Pakistan, Palau and Viet Nam. He has also supported the IFC and world bank on e-fuels and green hydrogen strategy development projects in Namibia and Pakistan.

With a background in industrial and specialty gases, including 27 years at BOC Gases, The BOC Group and Linde Gas, Stephen has intimate knowledge of e-fuels, hydrogen, ammonia and carbon dioxide from commercial, technical and operational perspectives. For 14 years, he was a global business leader in these FTSE100 and DAX30 companies.

As a member of the H2 View and **gasworld** editorial advisory boards, Stephen advises the direction for the leading hydrogen-focused international publications. Through H2 VIEW, World Hydrogen Leaders and Sustainable Aviation Futures, he has led Masterclasses covering many hydrogen, SAF and hydrogen derivatives themes in virtual and live sessions.

Stephen was session chair for the e-fuels and hydrogen propulsion track at the Bremen Hydrogen Technology Exhibition in September 2023 and chaired the same stream at that conference in Hamburg in 2024. He was also conference chair for the CO2 utilisation Summit in Hamburg in 2023 and the same event in Berlin in 2024.

